

Date Planned : / /	Daily Tutorial Sheet-3	Expected Duration : 30 Min
Actual Date of Attempt : / /	Level-1	Exact Duration :

31. The most reactive compound towards formation of cyanohydrin on treatment with HCN followed by acidification is:

(A) benzaldehyde (B) p-nitrobenzaldehyde

(C) p-methoxybenzaldehyde (D) p-hydroxybenzaldehyde

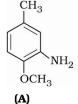
- **32**. Which factor(s) will increase the reactivity of > C = O group?
 - Presence of a group with positive inductive effect. I.
 - П. Presence of a group with negative inductive effect.
 - III. Presence of large alkyl group.

(A) Only I

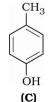
Only II (B)

(C) I and III (D) II and III

33. An organic compound reacts with chloroform in alkaline medium to give the compound A which adds hydrogen cyanide to form the compound B. The latter on acidic hydrolysis gives chiral carboxylic acid. The structure of the organic compound is:



(B)





34. The increasing order of the rate of HCN addition to compound A-D is:

> НСНО (a)

(A)

(b) CH₃COCH₃

d < b < c < a

(c) PhCOCH₃

(C)

(d)

(D)

PhCOPh

c < d < b < a

35. In the reaction, $X + NH_3 \longrightarrow (CH_3CHNH)_3$; X is:

(B)

(A) acetaldehyde ammonia trimer

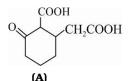
a < b < c < d

(B) para-formaldehyde

d < c < b < a

(C) formaldehyde (D) acetaldehyde

*36. The compound that undergoes decarboxylation readily by heating:









37. An organic compound 'A' burns with a sooty flame. It shows positive iodoform test. It does not answer Tollen's test. The compound A is:

(A) Acetophenone **(B)** Acetone

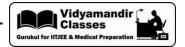
Salicyclic acid (C)

(D) Benzaldehyde

*38. In the following reduction, reducing agent used is:

(A) Na / EtOH (B) LiAlH₄ (C) NaAlH₄ (D)

H2/Pd-BaSO4



39. Identify A in the following sequence of reactions :

$$\begin{array}{c} \text{O} \\ \text{II} \\ \text{C}_{3}\text{H}_{5}\text{N(A)} \xrightarrow{\text{H}_{3}\text{O}^{+}} \xrightarrow{\text{PCl}_{5}} \xrightarrow{\text{(CH}_{3})_{2}\text{CuLi}} \xrightarrow{\text{CH}_{3}\text{CH}_{2}\text{C}\text{CH}_{3}} \\ \end{array}$$

(A) CH₃CH₂CN

(B) CH₃CH₂NC

(C) $CH_2 = CHCH = NH$

- (D) None of these
- **40.** Propanal reacts with methanamine in presence of H⁺ ions forming:
 - (A) anti-imine

(B) syn-imine

(C) Both (A) and (B)

- **(D)** None of the above
- *41. Which of the following pairs of reactants can form an enamine?

O || (A)
$$CH_3CH_2CH + [(CH_3)_2CH]_2NH$$

(B)
$$(CH_3)_3CCH + (CH_3)_2NH$$

(C)
$$+ (CH_3)_2NH$$

- (D) $CH_3 CHO + CH_3NH_2$
- **42.** An organic compound X gives a red precipitate on heating with Fehling's solution. Which one of the following reactions yields X as a major product?
 - (A) HCHO $\xrightarrow{\text{(i) CH}_3\text{MgI}}$ $\xrightarrow{\text{(ii) H}_2\text{O}}$
- **(B)** $C_2H_5Br + AgOH \xrightarrow{\Delta}$
- (C) $2C_2H_5Br + Ag_2O \xrightarrow{\Delta}$
- (D) $C_2H_2 + H_2O \xrightarrow{40\% \text{ H}_2SO_4} 1\% \text{ HgSO}_4 \to 60^{\circ}\text{ C}$
- **43.** Benzaldehyde and acetone can be best distinguished using:
 - (A) Fehling's solution

(B) sodium hydroxide solution

(C) 2, 4-DNP

- **(D)** Tollen's reagent
- *44. Identify the reaction which is used to obtain β hydroxy carbonyl compounds :
 - (A) Condensation reaction
- **(B)** Aldol condensation
- (C) Claisen-Schmidt reaction
- (D) Cannizzaro reaction
- **45.** An aromatic compound 'X' with molecular formula $C_9H_{10}O$ gives the following chemical tests.
 - **I.** Forms 2, 4-DNP derivative
 - II. Reduces Tollen's reagent
 - III. Undergoes Cannizzaro reaction and
 - **IV.** On vigorous oxidation 1, 2-benzenedicarboxylic acid is obtained. X is:



$$CHC$$
 C_2H_5



(A)

(B)

(C)

(D)